Organic Chemistry Example Problems

- **1.** An alkane is:
- a) a hydrocarbon with no double or triple bonds
- b) a hydrocarbon with only single or double bonds
- c) a hydrocarbon with a ring structure
- d) a hydrocarbon that has triple bonds

2. Which of the following compounds has a chiral carbon?







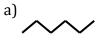


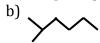


3. Which class of hydrocarbons can have cis-/trans- isomers:

- a) alkanes
- b) alkenes
- c) alkynes
- d) aromatics

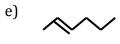
4. Which of the following compounds is an isomer of n-hexane?











5. What is the reaction product of $HCl + CH_2 = CH(CH_3)$?

- a) CH₂ClCH₂CH₃
- b) CH₃CHClCH₃
- c) $CH_2=CCl(CH_3)$
- d) ClCH=CH(CH₃)

6. Give the name for the compound: CH₃CH₂CHCH₃

CH₂CH₃

- a) 3-ethylbutane b) 2-ethylbutane c) 3-methylpentane
- d) 2-ethanyltetraane

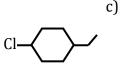
f) 3-methylquintane

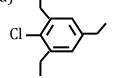
7. Which carbon skeleton structure represents 1-chloro-3ethylbenzene?

a)



b)







8. Which of the following alkenes has no cis-/trans- isomers?

- a) CHCl=CHCl
- b) CHCl=CH(CH₃) c) CCl₂=CH(CH₃) d) C(CH₃)H=C(CH₃)Cl
- e) CHBr=CHCl

9. Hydrogenation of CH₃(CH₂)₅CH=CH(CH₂)₃CH₃ (cis- isomer) is expected to result in a product

- a) is more polar b) is more volatile c) is less stable d) melts at a higher temperature

- e) has identical properties of the reactant

Organic Chemistry Example Problems

10. CH₃CH₂OCH₂CH₂CH₃ is an example of a/an: a) alcohol b) ether c) amine

e) ketone