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|---|---|
| 1. $\text{HClO}_{4(\text{aq})}$ | -- perchloric acid |
| 2. carbonic acid | -- $\text{H}_2\text{CO}_{3(\text{aq})}$ |
| 3. $\text{HIO}_{2(\text{aq})}$ | -- iodosic acid |
| 4. acetic acid | -- $\text{HC}_2\text{H}_3\text{O}_{2(\text{aq})}$ |
| 5. $\text{H}_2\text{CrO}_{4(\text{aq})}$ | -- chromic acid |
| 6. $\text{HNO}_{2(\text{aq})}$ | -- nitrous acid |
| 7. permanganic acid | -- $\text{HMnO}_{4(\text{aq})}$ |
| 8. $\text{HF}_{(\text{aq})}$ | -- hydrofluoric acid |
| 9. phosphorous acid | -- $\text{H}_3\text{PO}_{3(\text{aq})}$ |
| 10. $\text{HBr}_{(\text{aq})}$ | -- hydrobromic acid |
| 11. hydroiodic acid | -- $\text{HI}_{(\text{aq})}$ |
| 12. $\text{H}_3\text{PO}_{4(\text{aq})}$ | -- phosphoric acid |
| 13. nitric acid | -- $\text{HNO}_{3(\text{aq})}$ |
| 14. $\text{H}_2\text{S}_{(\text{aq})}$ | -- hydrosulfuric acid |
| 15. chlorous acid | -- $\text{HClO}_{2(\text{aq})}$ |
| 16. $\text{HIO}_{3(\text{aq})}$ | -- iodic acid |
| 17. sulfurous acid | -- $\text{H}_2\text{SO}_{3(\text{aq})}$ |
| 18. $\text{HBrO}_{2(\text{aq})}$ | -- bromous acid |
| 19. hydrocyanic acid | -- $\text{HCN}_{(\text{aq})}$ |
| 20. hypoiodous acid | -- $\text{HIO}_{(\text{aq})}$ |
| 21. $\text{HBrO}_{(\text{aq})}$ | -- hypobromous acid |
| 22. $\text{HClO}_{3(\text{aq})}$ | -- chloric acid |
| 23. $\text{H}_2\text{Se}_{(\text{aq})}$ | -- hydroselenic acid |
| 24. hydrochloric acid | -- $\text{HCl}_{(\text{aq})}$ |
| 25. perbromic acid | -- $\text{HBrO}_{4(\text{aq})}$ |
| 26. selenic acid | -- $\text{H}_2\text{SeO}_{4(\text{aq})}$ |
| 27. $\text{H}_2\text{C}_2\text{O}_{4(\text{aq})}$ | -- oxalic acid |
| 28. sulfuric acid | -- $\text{H}_2\text{SO}_{4(\text{aq})}$ |
| 29. $\text{HIO}_{4(\text{aq})}$ | -- periodic acid |
| 30. bromic acid | -- $\text{HBrO}_{3(\text{aq})}$ |
| 31. hypochlorous acid | -- $\text{HClO}_{(\text{aq})}$ |