Exam 1	
CHEM	160A

Name		
October 6, 2003		

1)	(2 points) Name two structures that are present in all eukaryotic cells but not in prokaryotic cells.

2) (4 points) Explain how entropy drives the formation of micelles.

3) (8 points) List the four types of weak interactions among biomolecules and the strength of each one.

Type of Interaction Strength (kJ/mol) 1. 2. 3. 4.

- 4) (2 points) What is the approximate strength of a covalent bond? _____kJ/mol
- 5) (3 points) How does K_w differ from K_a for water?

- 6) (1 point) Which has a higher pK_a?
 - a) A strong acid
 - b) A weak acid
- 7) (2 points) How is pK_a related to the pH scale?
- 8) (12 points) Phosphoric acid has three pK_as: 2.12, 7.21, and 12.32. You have available four reagents:

Liquid H₃PO₄ (18.1 M) NaH₂PO₄ (MW 138.01 g/mol) Na₂HPO₄ (MW 141.98 g/mol) Na₃PO₄ (MW 141.98 g/mol)

To make 2.00 L of 0.100 M phosphate buffer, pH 7.40:

- a) Circle the reagent you would use as the weak acid (HA).
- b) Put a box around the reagent you would use as the conjugate base.
- c) Calculate how much acid and its conjugate base you would measure out. (Give your answer in grams or milliliters.)

9)	(4 points) Draw a titra		I with a pK_a of 9.0.	
	(1 point) Circle the bu	ffering region.		
10) (1				
10) (1 point) Which amino acid has a pK _a nearest physiological pH?				
	· 			
11)	(12 points) Fill in this	table:		
,	1 /			
	Amino acid	Charge at pH 1.5	Charge at pH 7.0	

Aspartic Acid

12) (15 points) Show the chemical reaction for the formation of the dipeptide val-thr from L-valine and L-threonine. (Draw all structures.)

13) (1 p	point) What general type of reaction forms a peptide bond?
14) The	pKas of valine are 2.29 and 9.72. The pKas of threonine are 2.09 and 9.10.
	(3 points) What is the pI of the dipeptide val-thr? (You MUST show your work for credit.)
b)	(3 points) Would the dipeptide val-thr make a good buffer at pH 7.0? Why or why not?
15) (2]	point) How many peptide bonds are in a tripeptide?
, , .	points) Which of the following tripeptides would be retained the longest on a cation hange column at pH 7.0?
a)	ser-gly-thr
b)	asp-glu-ser
c)	gly-thr-lys
d)	glu-thr-lys
filtr hav	points) You attempt to separate five proteins from each other using a size exclusion (gel ration) column, with a resin that has a molecular weight cut-off of 25kD. The proteins e the following molecular weights (in kD): 3, 12, 22, 40, and 100. List the order of ion of these proteins from this column, starting with the proteins that elute first.

	points) Describe the similarities and o	differences between denaturing gel electrophoresis phy.
	points) How does information about t derstanding of a protein's function?	the sequence of a protein contribute to an
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	points) What is the sequence of the potential points in these fragments:	eptide that, when cleaved with the following
	psin	Trypsin
a)	FQRLV	a) NDEQFQR
b)	YCA	b) LVWMEK
c)	WMEKG	c) GYCA
d)	NDEQ	