Verschränkung versus Stosszahlansatz: The second law rests on low correlation levels in our cosmic neighborhood

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M. Hossein Partovi

Department of Physics & Astronomy, California State University, Sacramento

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Hossein Partovi (Sacramento State) Verschränkung vs. Stosszahlansatz

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### Abstract

The extremes of correlation level in the environment are shown to lead to opposite directions of the thermodynamic arrow. In a low-entropy environment where the mean correlation level is low, the **Stosszahlansatz scenario**, the Clausius inequality and the second law are established using general arguments of quantum information theory. For a high-correlation environment, the Verschränkung scenario, a model of entangled systems is constructed in which heat flows from the cold to the hot body in direct violation of the second law. This model clearly shows that the second law of thermodynamics is a consequence of quantum statistical dynamics under conditions of low ambient correlations and breaks down under opposite conditions. When the Stosszahlansatz scenario prevails, the second law is valid for systems of any size, although its utility is limited for microscopic systems where fluctuations are not negligible relative to mean values. While the von Neumann entropy and other information-theoretic objects are used extensively in establishing the above results, all conclusions are based on energy flow between systems without any assumption on the thermodynamic interpretation of such objects. The results of this work strongly support the expectation, first expressed by Boltzmann and subsequently elaborated by others, that the second law is an emergent consequence of fundamental dynamics operating in a low-entropy cosmological environment functioning as an ideal information sink.

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- I believe that the thermodynamic arrow and the second law and the phenomenon of reduction in quantum measurements are consequences of quantum dynamics as already known, and that methods of quantum information theory are best suited to their analysis and resolution.

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- An accelerating universe is an ideal entropy (waste information) dump, allowing the Earth to maintain a low-correlation environment and thermodynamics as we know it.
- We use methods of quantum information and base our conclusions on energy flow only, thereby avoiding controversial issues of interpretation, e.g., of entropic quantities.

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Let the systems be labeled i = 1, 2, ..., N, and the (von Neumann) entropies  $S^i$ . The two-body correlation information for the (i, j) pair is defined as  $I^{ij} = S^i + S^j - S^{ij}$ .

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Strong subadditivity implies  $S^i + S^j \leq S^{ik} + S^{jk}$ . There are N(N-1)/2 such inequalities, which, when aggregated, give

$$[N(N-1)/2]^{-1} \sum_{i< j=1}^{N(N-1)/2} I^{ij} \le N^{-1} \sum_{i=1}^{N} S^i.$$
(1)

Thus  $I_{av} \leq S_{av}$ , indicating that a small average entropy guarantees a low level of two-body correlations.

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Thus  $I_{av} \leq S_{av}$ , indicating that a small average entropy guarantees a low level of two-body correlations.

The significance of this result is the **universal** validity of Boltzmann's **Stosszahlansatz**, for all things large or small, as a likely condition in a low-entropy universe.

## The Fundamental Inequality

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Consider the evolution of any system initially in the equilibrium (Gibbs) state  $\rho_i = \exp(-\beta_i H_i)/Z_i$  to any state  $\rho_f$ , where  $\beta_i = 1/T_i$ ,  $H_i$ , and  $Z_i$  are the initial inverse temperature, Hamiltonian, and partition function respectively (Boltzmann's constant set to unity).

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In general, the Hamiltonian will change during the evolution (causing exchange of work), the system will evolve in a non-unitary manner, and  $\rho_f$  will be out of equilibrium.

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In general, the Hamiltonian will change during the evolution (causing exchange of work), the system will evolve in a non-unitary manner, and  $\rho_f$  will be out of equilibrium.

We can use the non-negativity of the relative entropy  $S(\rho_f \| \rho_i)$  to assert that

$$S(\rho_f \| \rho_i) = \beta_i \Delta U - \Delta S - \beta_i \operatorname{tr}(\rho_f \Delta H) \ge 0,$$
(2)

where  $\Delta U = U_f - U_i = \operatorname{tr}(\rho_f H_f) - \operatorname{tr}(\rho_i H_i)$ , and  $\Delta S = S_f - S_i$ . This is the fundamental inequality.

Rearranged, Eq. (2) reads

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For a small change of state, Eq. (3) reads

$$T_i dS \le dU + \delta W,\tag{4}$$

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# But why? Does it, or the fundamental inequality, violate time-reversal invariance?

$$S(\rho_i \| \rho_f) = -\beta_f \Delta U + \Delta S + \beta_f \operatorname{tr}(\rho_i \Delta H) \ge 0, \tag{6}$$

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Eqs. (5) and (7) are nevertheless consistent with equilibrium thermodynamics since, for any small process whose endpoints are equilibrium states, both (7) and (5) apply, implying that  $dS = \delta Q/T$ .

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Normal Thermodynamic Behavior

### Scenario S: normal thermodynamic behavior

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For a *Stosszahlansatz* universe, Scenario S, Eq. (1) guarantees low correlation levels, so a typical pair of systems A and B, in equilibrium at temperatures  $T^A$  and  $T^B$ , will be initially uncorrelated. Therefore  $\rho_i^{AB} = \rho_i^A \otimes \rho_i^B$  and  $S_i^{AB} = S_i^A + S_i^B$ .

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Upon interaction, the final state  $\rho_f^{AB}$  will in general be correlated so that  $S_f^{AB} \leq S_f^A + S_f^B$ .

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Upon interaction, the final state  $\rho_f^{AB}$  will in general be correlated so that  $S_f^{AB} \leq S_f^A + S_f^B$ . Since the pair is isolated,  $\rho_i^{AB}$  and  $\rho_f^{AB}$  are related unitarily and  $S_f^{AB} = S_i^{AB}$ .

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We therefore find

 $\Delta S^A + \Delta S^B \ge 0 \text{ (Scenario S: systems initially in equilibrium).}$ (8)

This equation is a consequence of universal *Stosszahlansatz*. It is the origin of irreversibility in the second law. Recall that the two systems were initially in equilibrium and only exchanged heat. Therefore, we find from Eq. (2) that  $\Delta S^A \leq Q^A/T^A$  and  $\Delta S^B \leq Q^B/T^B$ .

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Combining the latter with  $\Delta S^A + \Delta S^B \ge 0$ , we find  $Q^A/T^A + Q^B/T^B \ge 0$ , and using  $Q^A + Q^B = 0$  (isolated pair exchanging heat only), we conclude that  $Q^A(1/T^A - 1/T^B) \ge 0$ .

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Thus  $Q^A$  has the same sign as  $T^B - T^A$ , i.e., heat flows from the *initially* hotter system to the *initially* colder one, whence the normal thermodynamic arrow.

Note: Final states are in general not in equilibrium.

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We will now establish the **second law** in full generality by proving **Clausius' inequality** under Scenario S, the state of our universe. Consider a system S that undergoes a cyclic process in thermal contact with a series of heat reservoirs,  $R_j$  at temperature  $T_j^R$ , absorbing  $Q_j^S$  from the *j*th reservoir, and exchanging work as a result of changes in its Hamiltonian.

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The *j*th process starts with the state  $\rho_j^S \rho_j^R$  (Scenario S), where  $\rho_j^R$  is a Gibbs state at temperature  $T_j^R$  and  $\rho_j^S$  is arbitrary. The end state is  $\rho_j^{SR}$ .

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Just as in case S above, we find  $\Delta S_j^S + \Delta S_j^R \ge 0$ ,  $V_j^S$  notwithstanding, and applying inequality (2) to the *reservoir*, we conclude that  $\beta_j^R Q_j^R \ge \Delta S_j^R$ , where  $\beta_j^R = 1/kT_j^R$  refers to the initial temperature of the reservoir.

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Energy exchange between the system and the reservoir is subject to  $Q_j^R = -Q_j^S$  since work exchange does not involve the reservoir. Combining the last two inequalities, we find  $\beta_j^R Q_j^S + \Delta S_j^S \leq 0$ .

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$$\sum_{j} \beta_j^R Q_j^S \le 0. \tag{9}$$

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This is the Clausius inequality, which is equivalent to the second law. It is a theorem in quantum mechanics for a low-entropy universe.

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Note that the inverse temperature  $\beta_j^R$  refers to the reservoirs, not the system, as in general the system is not at equilibrium as it goes through the cycle. Note also that, as promised, our end results are statement about energy flow and do not involve entropic quantities (which have caused endless arguments in the literature).

Above, we derived the Clausius inequality on the basis of the condition  $\Delta S^A + \Delta S^B \ge 0$  which prevails when the systems A and B are initially uncorrelated. What if they are initially correlated?

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$$|\Omega^{AB}\rangle = Z^{-1/2} \sum_{i} \exp(-\gamma \epsilon_i/2) |i;A\rangle |i;B\rangle, \tag{10}$$

where  $|i; A\rangle$  ( $|i; B\rangle$ ) is the *i*th energy eigenvector of system A (B),  $\gamma$  is a positive number, and  $Z^{-1/2}$  is a normalization constant.

Reversed Thermodynamic Behavior

The individual states of the two systems, which are the marginal states of  $|\Omega^{AB}\rangle\langle\Omega^{AB}|$ , are easily found to be thermal equilibrium (Gibbs) states at temperatures  $T^A = 1/(\gamma \mu^A)$  and  $T^B = 1/(\gamma \mu^B)$ , respectively.

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This reversal leads to  $Q^A/T^A + Q^B/T^B \ge \Delta S^A + \Delta S^B$ , which allows both directions of heat flow, including that from the initially colder body to the hotter one.

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A detailed dilute gas model exhibiting this reversal explicitly is constructed in *Phys. Rev. E* **77**, 021110 (2008). How do we understand this bizarre behavior? The clue is in the dual character of  $|\Omega^{AB}\rangle$ : (i) its purity forces  $S^A$  and  $S^B$  to move in lock-step,

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Pre-existing correlations can neutralize the overwhelming statistical biases that normally underlie the second law.

Scenarios S and V are paradigms, informing us that the second law is not an independent, or inviolable, law of nature. Rather, it is a consequence of quantum mechanics conditioned on a low-correlation environment, and breaks down under the opposite conditions.

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- These laws hold for all systems great and small. However, it is only for meso- and macro-sytems that fluctuations are small enough to make thermodynamic quantities useful.
- The reversibility issues that so haunted Boltzmann at the end of the nineteenth century are seen to arise from the initial conditions prevailing in the environment. To his credit, Boltzmann had already guessed that!
- Understanding thermodynamic behavior as a consequence of quantum dynamics conditioned on the environment is a prerequisite to the resolution of the reduction phenomenon in quantum measurements.

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### References

This talk is for the most part based on

- M. Hossein Partovi, "Entanglement versus *Stosszahlansatz*: Disappearance of the thermodynamic arrow in a high-correlation environment," *Phys. Rev. E* **77**, 021110 (2008).
- M. Hossein Partovi, "Quantum Thermodynamics," *Phys. Letters* A **137**, 440 (1989).
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Further results extending those of the first paper above can be found in

• David Jennings and Terry Rudolph, "Entanglement and the thermodynamic arrow of time," *Phys. Rev. E* **81**, 061130 (2010).

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