Report: BS/BA Chemistry

Question 1: Program Learning Outcomes

Q1.1. Which of the following Program Learning Outcomes (PLOs) and Sac State Baccalaureate Learning Goals (BLGs) did you assess? [Check all that apply]

- 1. Critical Thinking
- 2. Information Literacy
- 3. Written Communication
- 4. Oral Communication
- 5. Quantitative Literacy
- 6. Inquiry and Analysis
- 7. Creative Thinking
- 8. Reading
- 9. Team Work
- 10. Problem Solving
- 11. Civic Knowledge and Engagement
- 12. Intercultural Knowledge and Competency
- 13. Ethical Reasoning
- 14. Foundations and Skills for Lifelong Learning
- 15. Global Learning
- 16. Integrative and Applied Learning
- 17. Overall Competencies for GE Knowledge
- 18. Overall Competencies in the Major/Discipline
- **19. Other, specify any assessed PLOs not included above:**
  a. Laboratory skills
  b. 
  c. 

Q1.2. Please provide more detailed background information about EACH PLO you checked above and other information such as how your specific PLOs are explicitly linked to the Sac State BLGs:

<table>
<thead>
<tr>
<th>Learning Outcome</th>
<th>Courses assessed</th>
<th>Year and frequency of assessment</th>
<th>Assessment activity</th>
<th>Evaluation</th>
<th>Data collection</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>A. Laboratory Knowledge and Skills</strong></td>
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<tr>
<td>1. the basic analytical and technical skills to work effectively in the various fields of chemistry</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
</tr>
<tr>
<td>2. the ability to perform accurate quantitative</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation</td>
</tr>
</tbody>
</table>

For instructions and guidelines visit our [website](#) or [contact us](#) for more help.
<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Title</th>
<th>Rubric</th>
<th>Evaluation Method</th>
<th>Faculty Evaluation</th>
<th>Department Chair Evaluation</th>
<th>Office of Assessment Evaluation</th>
</tr>
</thead>
<tbody>
<tr>
<td>-</td>
<td>A. Scientific Measurements with an understanding of the theory and use of contemporary chemical instrumentation, interpret experimental results, perform calculations on these results and draw reasonable, accurate conclusions.</td>
<td>Not assessed at program level</td>
<td>Multiple faculty evaluation during department poster session</td>
<td>N/A</td>
<td>N/A</td>
<td>N/A</td>
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<tr>
<td>3. the ability to synthesize, separate and characterize compounds using published reactions, protocols, standard laboratory equipment, and modern instrumentation.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
<td>N/A</td>
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<tr>
<td>4. the ability to use information technology tools such as the Internet and computer-based literature searches as well as printed literature resources to locate and retrieve scientific information needed for laboratory or theoretical work.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
<td>N/A</td>
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<tr>
<td>5. the ability to present scientific and technical information resulting from laboratory experimentation in both written and oral formats.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
<td>N/A</td>
</tr>
<tr>
<td>6. knowledge and understanding of the issues of safety regulations in the use of chemicals in their laboratory work.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Not assessed at program level</td>
<td>N/A</td>
<td>N/A</td>
<td>N/A</td>
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<tr>
<td>B. Computer, Library and Information Skills</td>
<td>1. the ability to make effective use of the library and other information resources in chemistry, including the primary literature, tabulated data, and secondary sources such as the Internet.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
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<td>2. the ability to make effective use of computers in chemistry applications</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
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<tr>
<td>C. Oral and Written Communication Skills in Chemistry</td>
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<tr>
<td>1. adequate skills in technical writing and oral presentations.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
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<tr>
<td>2. the ability to communicate scientific information in oral and written formats to both scientists and nonscientists.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
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</table>

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<thead>
<tr>
<th>D. Quantitative Reasoning Skills</th>
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<tbody>
<tr>
<td>1. ability to accurately collect and interpret numerical data.</td>
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<tr>
<td>2. ability to solve problems competently using extrapolation, approximation, precision, accuracy, rational estimation and statistical validity.</td>
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<tr>
<td>3. proficiency in the scientific method (formulating hypotheses and arriving at appropriate answers and conclusions)</td>
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</table>

<table>
<thead>
<tr>
<th>E. Knowledge of Chemical Principles and Facts</th>
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<td>c</td>
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</tbody>
</table>
1. a working knowledge of chemical principles appropriate to a chemistry degree program to include thermodynamics, equilibrium, kinetics, quantum mechanics, structures of materials, reactivities of substances, synthesis, isolation and identification of compounds.

2. a mastery of a broad set of factual chemical knowledge concerning the properties of substances, molecules, and atoms.

<table>
<thead>
<tr>
<th>Question</th>
<th>110, 24/124, 160A/B</th>
<th>Assessment conducted in 110, 124, 160B every semester offered</th>
<th>ACS Standardized Exam&lt;sup&gt;2&lt;/sup&gt;</th>
<th>ACS Exzm key</th>
<th>Course instructor</th>
</tr>
</thead>
<tbody>
<tr>
<td>Q1.2.1.</td>
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<td>Do you have rubrics for your PLOs?</td>
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<tr>
<td>1. Yes, for all PLOs</td>
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<td>2. Yes, but for some PLOs</td>
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<td>3. No rubrics for PLOs</td>
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<td>4. N/A</td>
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<tr>
<td>5. Other, specify:</td>
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</tbody>
</table>

Q1.3.
Are your PLOs closely aligned with the mission of the university?

| 1. Yes | 2. No | 3. Don't know |

Q1.4.
Is your program externally accredited (other than through WASC Senior College and University Commission (WSCUC))?

| 1. Yes | 2. No (skip to Q1.5) | 3. Don't know (skip to Q1.5) |

Q1.4.1.
If the answer to Q1.4 is yes, are your PLOs closely aligned with the mission/goals/outcomes of the accreditation agency?

| 1. Yes | 2. No |

**ACS Standardized Exam<sup>2</sup>**

<table>
<thead>
<tr>
<th>Course instructor</th>
</tr>
</thead>
</table>
3. Don’t know

**Q1.5.**
Did your program use the Degree Qualification Profile (DQP) to develop your PLO(s)?
- 1. Yes
- 2. No, but I know what the DQP is
- 3. No, I don’t know what the DQP is
- 4. Don’t know

**Q1.6.**
Did you use action verbs to make each PLO measurable?
- 1. Yes
- 2. No
- 3. Don’t know

(Remember: Save your progress)

Question 2: Standard of Performance for the Selected PLO

**Q2.1.**
Select ONE(1) PLO here as an example to illustrate how you conducted assessment (be sure you checked the correct box for this PLO in Q1.1):

Overall Competencies in the Major/Discipline

**Q2.1.1.**
Please provide more background information about the specific PLO you’ve chosen in Q2.1.

We administer the American Chemical Society content area exams in three courses, Chem 124, Chem 160B, and Chem 110. We compare student performance to national norms.

**Q2.2.**
Has the program developed or adopted explicit standards of performance for this PLO?
- 1. Yes
- 2. No
- 3. Don’t know
- 4. N/A

**Q2.3.**
Please provide the rubric(s) and standards of performance that you have developed for this PLO here or in the appendix.

The exam is scored using ACS exam keys. The implicit standard of performance is to be at or above the national norm.

Note: ACS exam keys are confidential documents that we cannot disseminate.

No file attached

**Q2.4. PLO | Q2.5. Stdrd | Q2.6. Rubric**
Please indicate where you have published the PLO, the standard of performance, and the rubric that was used to measure the PLO:
1. In SOME course syllabi/assignments in the program that address the PLO
2. In ALL course syllabi/assignments in the program that address the PLO
3. In the student handbook/advising handbook
4. In the university catalogue
5. On the academic unit website or in newsletters
6. In the assessment or program review reports, plans, resources, or activities
7. In new course proposal forms in the department/college/university
8. In the department/college/university’s strategic plans and other planning documents
9. In the department/college/university’s budget plans and other resource allocation documents
10. Other, specify: 

Question 3: Data Collection Methods and Evaluation of Data Quality for the Selected PLO

Q3.1.
Was assessment data/evidence collected for the selected PLO?
  1. Yes
  2. No (skip to Q6)
  3. Don't know (skip to Q6)
  4. N/A (skip to Q6)

Q3.1.1.
How many assessment tools/methods/measures in total did you use to assess this PLO?
1

Q3.2.
Was the data scored/evaluated for this PLO?
  1. Yes
  2. No (skip to Q6)
  3. Don't know (skip to Q6)
  4. N/A (skip to Q6)

Q3.2.1.
Please describe how you collected the assessment data for the selected PLO. For example, in what course(s) or by what means were data collected:
Faculty submitted performance range, average and means for comparison to national norms for courses indicated above.

(Remember: Save your progress)

Question 3A: Direct Measures (key assignments, projects, portfolios, etc.)

Q3.3.
Were direct measures (key assignments, projects, portfolios, course work, student tests, etc.) used to assess this PLO?
  1. Yes
  2. No (skip to Q3.7)
3. Don't know (skip to Q3.7)

**Q3.3.1.**
Which of the following direct measures were used? [Check all that apply]

1. Capstone project (e.g. theses, senior theses), courses, or experiences
2. Key assignments from required classes in the program
3. Key assignments from elective classes
4. Classroom based performance assessment such as simulations, comprehensive exams, or critiques
5. External performance assessments such as internships or other community-based projects
6. E-Portfolios
7. Other Portfolios
8. Other, specify: __________________________

**Q3.3.2.**
Please explain and attach the direct measure you used to collect data:

See Q 2.3  We cannot disseminate ACS exam materials.

[No file attached] [No file attached]

**Q3.4.**
What tool was used to evaluate the data?

1. No rubric is used to interpret the evidence (skip to Q3.4.4.)
2. Used rubric developed/modified by the faculty who teaches the class (skip to Q3.4.2.)
3. Used rubric developed/modified by a group of faculty (skip to Q3.4.2.)
4. Used rubric pilot-tested and refined by a group of faculty (skip to Q3.4.2.)
5. The VALUE rubric(s) (skip to Q3.4.2.)
6. Modified VALUE rubric(s) (skip to Q3.4.2.)
7. Used other means (Answer Q3.4.1.)

**Q3.4.1.**
If you used other means, which of the following measures was used? [Check all that apply]

1. National disciplinary exams or state/professional licensure exams (skip to Q3.4.4.)
2. General knowledge and skills measures (e.g. CLA, ETS PP, etc.) (skip to Q3.4.4.)
3. Other standardized knowledge and skill exams (e.g. ETC, GRE, etc.) (skip to Q3.4.4.)
4. Other, specify: __________________________ (skip to Q3.4.4.)

**Q3.4.2.**
Was the rubric aligned directly and explicitly with the PLO?

1. Yes
2. No
3. Don't know
4. N/A

**Q3.4.3.**
Was the direct measure (e.g. assignment, thesis, etc.) aligned directly and explicitly with the rubric?

1. Yes
Q3.4.4. Was the direct measure (e.g. assignment, thesis, etc.) aligned directly and explicitly with the PLO?
1. Yes
2. No
3. Don't know
4. N/A

Q3.5. How many faculty members participated in planning the assessment data collection of the selected PLO?

three

Q3.5.1. How many faculty members participated in the evaluation of the assessment data for the selected PLO?

one

Q3.5.2. If the data was evaluated by multiple scorers, was there a norming process (a procedure to make sure everyone was scoring similarly)?
1. Yes
2. No
3. Don't know
4. N/A

Q3.6. How did you select the sample of student work (papers, projects, portfolios, etc.)?

All student exams are scored.

Q3.6.1. How did you decide how many samples of student work to review?

All students take the exam.

Q3.6.2. How many students were in the class or program?
Q3.6.3. How many samples of student work did you evaluate?
All students who took the exam were evaluated.

Q3.6.4. Was the sample size of student work for the direct measure adequate?
- 1. Yes
- 2. No
- 3. Don't know

(Remember: Save your progress)

Question 3B: Indirect Measures (surveys, focus groups, interviews, etc.)

Q3.7. Were indirect measures used to assess the PLO?
- 1. Yes
- 2. No (skip to Q3.8)
- 3. Don't Know (skip to Q3.8)

Q3.7.1. Which of the following indirect measures were used? [Check all that apply]
- 1. National student surveys (e.g. NSSE)
- 2. University conducted student surveys (e.g. OIR)
- 3. College/department/program student surveys or focus groups
- 4. Alumni surveys, focus groups, or interviews
- 5. Employer surveys, focus groups, or interviews
- 6. Advisory board surveys, focus groups, or interviews
- 7. Other, specify:

Q3.7.1.1. Please explain and attach the indirect measure you used to collect data:

No file attached

Q3.7.2. If surveys were used, how was the sample size decided?
Q3.7.3. If surveys were used, how did you select your sample:

Q3.7.4. If surveys were used, what was the response rate?

Question 3C: Other Measures (external benchmarking, licensing exams, standardized tests, etc.)

Q3.8. Were external benchmarking data, such as licensing exams or standardized tests, used to assess the PLO?

- 1. Yes
- 2. No (skip to Q3.8.2)
- 3. Don't Know (skip to Q3.8.2)

Q3.8.1. Which of the following measures was used? [Check all that apply]

- 1. National disciplinary exams or state/professional licensure exams
- 2. General knowledge and skills measures (e.g. CLA, ETS PP, etc.)
- 3. Other standardized knowledge and skill exams (e.g. ETC, GRE, etc.)
- 4. Other, specify:

Q3.8.2. Were other measures used to assess the PLO?

- 1. Yes
- 2. No (skip to Q4.1)
- 3. Don't know (skip to Q4.1)

Q3.8.3. If other measures were used, please specify:
Question 4: Data, Findings, and Conclusions

Q4.1.
Please provide simple tables and/or graphs to summarize the assessment data, findings, and conclusions for the selected PLO for Q2.1:

Data are similar to past years with students scoring near the national norm. Also typical is the higher performance in Chem 110. We believe this is due to the experience of students in this culminating experience class.

2015-2016 ACS exam results

<table>
<thead>
<tr>
<th>Course</th>
<th>Area</th>
<th>Class level</th>
<th># sections</th>
<th>Mean</th>
<th>Percentile</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chem 124</td>
<td>Organic chem</td>
<td>Soph/Jr</td>
<td>4</td>
<td>36</td>
<td>51</td>
</tr>
<tr>
<td>Chem 110</td>
<td>Inorganic chem</td>
<td>Mostly Sr</td>
<td>1</td>
<td>34</td>
<td>58</td>
</tr>
<tr>
<td>Chem 160B</td>
<td>Biochem</td>
<td>Jr/Sr</td>
<td>1</td>
<td>33</td>
<td>53</td>
</tr>
</tbody>
</table>

Q4.2.
Are students doing well and meeting the program standard? If not, how will the program work to improve student performance of the selected PLO?

We expect at least the national average and that is what we are seeing.

Q4.3.
For the selected PLO, the student performance:

1. Exceeded expectation/standard
2. Met expectation/standard
3. Partially met expectation/standard
4. Did not meet expectation/standard
5. No expectation/standard has been specified
6. Don't know

Question 4A: Alignment and Quality

Q4.4.
Did the data, including the direct measures, from all the different assessment tools/measures/methods directly align with the PLO?

1. Yes
2. No
Q4.5.
Were all the assessment tools/measures/methods that were used good measures of the PLO?

- 1. Yes
- 2. No
- 3. Don't know

Question 5: Use of Assessment Data (Closing the Loop)

Q5.1.
As a result of the assessment effort and based on prior feedback from OAPA, do you anticipate making any changes for your program (e.g. course structure, course content, or modification of PLOs)?

- 1. Yes
- 2. No (skip to Q5.2)
- 3. Don't know (skip to Q5.2)

Q5.1.1.
Please describe what changes you plan to make in your program as a result of your assessment of this PLO. Include a description of how you plan to assess the impact of these changes.

Q5.1.2.
Do you have a plan to assess the impact of the changes that you anticipate making?

- 1. Yes
- 2. No
- 3. Don't know

Q5.2.
How have the assessment data from the last annual assessment been used so far? [Check all that apply]

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<tr>
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<tbody>
<tr>
<td>1. Improving specific courses</td>
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<tr>
<td>2. Modifying curriculum</td>
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<tr>
<td>3. Improving advising and mentoring</td>
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<td>4. Revising learning outcomes/goals</td>
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<td>5. Revising rubrics and/or expectations</td>
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<tr>
<td>6. Developing/updating assessment plan</td>
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<tr>
<td>7. Annual assessment reports</td>
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<td>8. Program review</td>
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<td>9. Prospective student and family information</td>
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<td>10. Alumni communication</td>
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<td>11. WSCUC accreditation (regional accreditation)</td>
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<td>12. Program accreditation</td>
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<td>13. External accountability reporting requirement</td>
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</table>
14. Trustee/Governing Board deliberations
15. Strategic planning
16. Institutional benchmarking
17. Academic policy development or modifications
18. Institutional improvement
19. Resource allocation and budgeting
20. New faculty hiring
21. Professional development for faculty and staff
22. Recruitment of new students

23. Other, specify:

Q5.2.1.
Please provide a detailed example of how you used the assessment data above:

We use the ACS exams as a benchmark of content knowledge.

(Remember: Save your progress)

Additional Assessment Activities

Q6.
Many academic units have collected assessment data on aspect of their program that are not related to the PLOs (i.e. impacts of an advising center, etc.). If your program/academic unit has collected data on program elements, please briefly report your results here:

Q7.
What PLO(s) do you plan to assess next year? [Check all that apply]

- 1. Critical Thinking
- 2. Information Literacy
- 3. Written Communication
- 4. Oral Communication
- 5. Quantitative Literacy
- 6. Inquiry and Analysis
- 7. Creative Thinking
- 8. Reading
- 9. Team Work
- 10. Problem Solving
- 11. Civic Knowledge and Engagement
12. Intercultural Knowledge and Competency
13. Ethical Reasoning
14. Foundations and Skills for Lifelong Learning
15. Global Learning
16. Integrative and Applied Learning
17. Overall Competencies for GE Knowledge
18. Overall Competencies in the Major/Discipline
19. Other, specify any PLOs not included above:
   a. Lab skills
   b. 
   c. 

Q8. Please attach any additional files here:
   No file attached

Q8.1. Have you attached any files to this form? If yes, please list every attached file here:

Program Information (Required)

P1.
Program/Concentration Name(s): [by degree]
BS/BA Chemistry

P1.1.
Program/Concentration Name(s): [by department]
Chemistry BS/BA

P2.
Report Author(s):
Linda Roberts

P2.1.
Department Chair/Program Director:
Linda Roberts

P2.2.
Assessment Coordinator:

P3.
Department/Division/Program of Academic Unit
Chemistry

P4.
College:
College of Natural Science & Mathematics

P5.
Total enrollment for Academic Unit during assessment semester (see Departmental Fact Book):
P6. Program Type:
1. Undergraduate baccalaureate major
2. Credential
3. Master's Degree
5. Other, specify:

P7. Number of undergraduate degree programs the academic unit has?
5

P7.1. List all the names:
Chemistry BS
Biochemistry BS
Chemistry BA
Chemistry BA, concentration Biochemistry
Chemistry BA, concentration Forensics

P7.2. How many concentrations appear on the diploma for this undergraduate program?
2

P8. Number of master's degree programs the academic unit has?
2

P8.1. List all the names:
Master of Science in Chemistry
Master of Science in Chemistry, Biochemistry concentration

P8.2. How many concentrations appear on the diploma for this master's program?
1

P9. Number of credential programs the academic unit has?
0

P9.1. List all the names:
P10. Number of **doctorate degree programs** the academic unit has?
0

P10.1. List all the names:

<table>
<thead>
<tr>
<th>When was your assessment plan developed?</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Before 2010-11</td>
</tr>
<tr>
<td>2. 2011-12</td>
</tr>
<tr>
<td>3. 2012-13</td>
</tr>
<tr>
<td>4. 2013-14</td>
</tr>
<tr>
<td>5. 2014-15</td>
</tr>
<tr>
<td>6. No Plan</td>
</tr>
<tr>
<td>7. Don't know</td>
</tr>
</tbody>
</table>

P11. last updated?

P11.3. Please attach your latest assessment plan:

![Chemistryassessmentplan.pdf 283.32 KB](attachment:Chemistryassessmentplan.pdf)

P12. Has your program developed a **curriculum map**?

1. Yes
2. No
3. Don't know

P12.1. Please attach your latest curriculum map:

![Chemistryassessmentplan.pdf 283.32 KB](attachment:Chemistryassessmentplan.pdf)

P13. Has your program indicated in the curriculum map where assessment of **student learning** occurs?

1. Yes
2. No
3. Don't know

P14. Does your program have a capstone class?

1. Yes, indicate: Chem 141, 110L, 164 (multiple due to different degree plans)
2. No
3. Don't know

P14.1. Does your program have any capstone project?

1. Yes
2. No
3. Don't know
(Remember: Save your progress)
Chemistry Department Policy
Assessment: Undergraduate Programs

1. MISSION STATEMENT
The Chemistry Department offers academic programs which provide students with a liberal arts background and the theoretical knowledge and experimental skills to prepare for entry-level careers in the public and private sectors involving chemistry, to enter a graduate school in chemistry or to qualify for the credential program. The curriculum is organized into two primary degree programs: Bachelor of Arts and Bachelor of Science. The Bachelor of Arts degree has three concentrations: General, Biochemistry and Forensics. Each degree program has a core set of lower division and upper division courses that provide fundamental knowledge and experimental training; upper division elective courses expand upon the fundamental information learned in the core courses. Students are encouraged to participate in the life of the University and department and to work with a faculty member in a scholarly project.

2. SUMMER ADVISEMENT AND ORIENTATION PERIODS FOR NEWLY ADMITTED FRESHMEN AND TRANSFER STUDENTS
A. The student will take appropriate diagnostic placement exams in English and Mathematics administered by the University
B. Chemistry faculty advisors will meet with students to assess appropriate placement in Chemistry courses.
C. Students planning to take Chemistry 1A are required to pass a diagnostic placement examination. If a passing score is not achieved, students are advised to enroll in Chemistry 4.

3. FRESHMEN THROUGH SENIOR YEARS
A. Students not taking the Chemistry 1A diagnostic test before classes start will take it in the first laboratory period.
B. Students must take the appropriate sequence of English, Mathematics and Physics courses in order to succeed as a chemistry major. Progress in these areas will be monitored by Chemistry faculty advisors and appropriate recommendations will be given to students based on their performance in the prerequisite courses.
C. Students are expected to begin taking required English and Mathematics courses in General Education in the first semester of attendance or have completed them if they are transferring.
D. Students are expected to develop their competence in eight skill areas. Five of these are transferable skills taken from general education courses and three [the last three] are more specifically aimed at students majoring in Chemistry.

- Effective Writing
- Effective Oral Communication
- Critical Thinking (logic and rhetoric)
- Problem Solving and Critical Thinking
- Quantitative Analysis
- Library and Scientific Information Literacy
- Computer literacy
- Laboratory Skills

E. A capstone project is required of students in Chemistry 125, 141 or 164 to assess students’ ability to transfer many of the skills listed above to a selected laboratory project. The projects
will require the use of chemical literature, doing a laboratory project, preparing a paper, and presenting the project in a poster presentation format to students, faculty and others. Students will be expected to be able to answer questions orally. Faculty will make a holistic judgment of students’ ability to transfer liberal arts skills to the field of chemistry. Also, students will be assigned a grade.

4. BEYOND GRADUATION
A. Recent Alumni will be surveyed every three years to determine their satisfaction regarding the quality of their undergraduate education and their preparedness for graduate school and/or their entry level position in industry. They will also be asked to suggest changes for improvements in the curriculum.
B. All Alumni will be able to inform the Chemistry Department of their successes and concerns in an alumni questionnaire that will be included in any Chemistry Department newsletter.

5. BS CHEMISTRY CURRICULUM*

<table>
<thead>
<tr>
<th></th>
<th>FRESHMAN</th>
<th>SOPHOMORE</th>
<th>JUNIOR</th>
<th>SENIOR</th>
</tr>
</thead>
<tbody>
<tr>
<td>CRITICAL THINKING</td>
<td>GE:AREA A3</td>
<td>CHEM 24, 31, 124</td>
<td>CHEM 140A, 140B, 141</td>
<td>CHEM 110, 133, 198</td>
</tr>
<tr>
<td>EFFECTIVE WRITING</td>
<td>E: AREA A2</td>
<td>CHEM 25</td>
<td>CHEM 125, 141</td>
<td>CHEM 110L, 198</td>
</tr>
<tr>
<td>EFFECTIVE ORAL COMMUNICATION</td>
<td>GE:AREA A1</td>
<td></td>
<td>CHEM 125</td>
<td>CHEM 110L, 198</td>
</tr>
<tr>
<td>PROBLEM SOLVING AND CREATIVE THINKING</td>
<td>GE: AREA B4 CHEM 1A, 1B</td>
<td>GE: AREA B CHEM 24, 25, 31, 124</td>
<td>CHEM 125, 140A, 140B, 141</td>
<td>CHEM 110L, 198</td>
</tr>
<tr>
<td>QUANTITATIVE ANALYSIS</td>
<td>GE: AREA B4</td>
<td>GE: AREA B CHEM 24, 131</td>
<td>CHEM 125, CHEM 133, 198</td>
<td>CHEM 110L, 198</td>
</tr>
<tr>
<td>LIBRARY AND INFORMATION LITERACY</td>
<td>GE: AREA B CHEM 1A, 1B</td>
<td>CHEM 125, 141</td>
<td>CHEM 110L, 198</td>
<td></td>
</tr>
<tr>
<td>COMPUTER LITERACY</td>
<td>CHEM: 1B</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>LABORATORY SKILLS</td>
<td>GE: AREA B CHEM 1A, 1B</td>
<td>CHEM 25, 31</td>
<td>CHEM 125, 141</td>
<td>CHEM 110L, 133, 198</td>
</tr>
</tbody>
</table>

Note: Six units of elective courses are required and these also will fit into the above matrix.
*Core and required upper division courses are typically listed. Elective courses will add breadth to the assessment areas.

### BA, GENERAL CHEMISTRY, CURRICULUM

<table>
<thead>
<tr>
<th></th>
<th>FRESHMAN</th>
<th>SOPHOMORE</th>
<th>JUNIOR</th>
<th>SENIOR</th>
</tr>
</thead>
<tbody>
<tr>
<td>CRITICAL THINKING</td>
<td>GE: AREA A3</td>
<td>CHEM 24, 31, 124</td>
<td>CHEM 140A, 140B, 141</td>
<td></td>
</tr>
<tr>
<td>EFFECTIVE WRITING</td>
<td>GE: AREA A2</td>
<td>CHEM 25</td>
<td>CHEM 141</td>
<td>TWO ELECTIVE COURSES</td>
</tr>
<tr>
<td>EFFECTIVE ORAL COMMUNICATION</td>
<td>GE: AREA A1</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>PROBLEM SOLVING AND CREATIVE THINKING</td>
<td>GE: AREA B4</td>
<td>GE: AREA B</td>
<td>CHEM 140A, 140B</td>
<td>TWO ELECTIVE COURSES</td>
</tr>
<tr>
<td>QUANTITATIVE ANALYSIS</td>
<td>GE: AREA B4</td>
<td>GE: AREA B</td>
<td>CHEM 140A, 140B, 141</td>
<td>TWO ELECTIVE COURSES</td>
</tr>
<tr>
<td>LIBRARY AND INFORMATION LITERACY</td>
<td></td>
<td></td>
<td>CHEM 141</td>
<td></td>
</tr>
<tr>
<td>COMPUTER LITERACY</td>
<td>CHEM: 1B</td>
<td></td>
<td>CHEM 141</td>
<td></td>
</tr>
<tr>
<td>LABORATORY SKILLS</td>
<td>GE: AREA B</td>
<td>CHEM 25, 31</td>
<td>CHEM 141</td>
<td>TWO ELECTIVE COURSES</td>
</tr>
</tbody>
</table>

**Note:** 12 units of upper division chemistry elective courses, including two courses containing a laboratory, will also fit into the above matrix.

### BA, BIOCHEMISTRY, CURRICULUM

<table>
<thead>
<tr>
<th></th>
<th>FRESHMAN</th>
<th>SOPHOMORE</th>
<th>JUNIOR</th>
<th>SENIOR</th>
</tr>
</thead>
<tbody>
<tr>
<td>Course</td>
<td>FRESHMAN</td>
<td>SOPHOMORE</td>
<td>JUNIOR</td>
<td>SENIOR</td>
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<tr>
<td>----------------------------------------</td>
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</tr>
<tr>
<td>EFFECTIVE WRITING</td>
<td>GE: AREA A2 CHEM 1A, 1B</td>
<td>CHEM 25</td>
<td>CHEM 162</td>
<td>CHEM 164</td>
</tr>
<tr>
<td>EFFECTIVE ORAL COMMUNICATION</td>
<td>GE: AREA A1</td>
<td></td>
<td></td>
<td>CHEM 164</td>
</tr>
<tr>
<td>PROBLEM SOLVING AND CREATIVE THINKING</td>
<td>GE: AREA B4 CHEM 1A, 1B</td>
<td>GE: AREA B CHEM 24, 25, 31, 124</td>
<td>CHEM 142</td>
<td>CHEM 160A, 160B</td>
</tr>
<tr>
<td>QUANTITATIVE ANALYSIS</td>
<td>GE: AREA B4 CHEM 1A, 1B</td>
<td>GE: AREA B CHEM 31</td>
<td>CHEM 142,162</td>
<td>CHEM 164</td>
</tr>
<tr>
<td>LIBRARY AND INFORMATION LITERACY</td>
<td></td>
<td></td>
<td>CHEM 160B</td>
<td>CHEM 164</td>
</tr>
<tr>
<td>COMPUTER LITERACY</td>
<td>CHEM: 1B</td>
<td></td>
<td></td>
<td>CHEM 164</td>
</tr>
<tr>
<td>LABORATORY SKILLS</td>
<td>GE: AREA B CHEM 1A, 1B</td>
<td>CHEM 25, 31</td>
<td>CHEM 141</td>
<td>CHEM 162, 164</td>
</tr>
</tbody>
</table>

**Note:** Two upper division courses must also be chosen from a prescribed list of courses in the Biological Sciences Department. These will also fit into the above matrix.

**BA, FORENSIC CHEMISTRY, CURRICULUM**

<table>
<thead>
<tr>
<th>Course</th>
<th>FRESHMAN</th>
<th>SOPHOMORE</th>
<th>JUNIOR</th>
<th>SENIOR</th>
</tr>
</thead>
<tbody>
<tr>
<td>CRITICAL THINKING</td>
<td>GE: AREA A3 CHEM 1A, 1B</td>
<td>CHEM 24, 31, 124</td>
<td>CHEM 140A, 140B OR 142</td>
<td>Chem 110* Chem 164*</td>
</tr>
<tr>
<td>EFFECTIVE WRITING</td>
<td>GE: AREA A2 CHEM 1A, 1B</td>
<td>CHEM 25</td>
<td>CHEM 141*</td>
<td>Chem 110L* Chem 164*</td>
</tr>
<tr>
<td>EFFECTIVE ORAL COMMUNICATION</td>
<td>GE: AREA A1</td>
<td></td>
<td></td>
<td>Chem 110L* Chem 164*</td>
</tr>
<tr>
<td>PROBLEM SOLVING AND CREATIVE THINKING</td>
<td>GE: AREA B4 CHEM 1A, 1B</td>
<td>GE: AREA B CHEM 24, 25, 31, 124</td>
<td>CHEM 140A, 140B OR 142, 141*</td>
<td></td>
</tr>
<tr>
<td>QUANTITATIVE ANALYSIS</td>
<td>GE: AREA B4 CHEM 1A, 1B</td>
<td>GE: AREA B CHEM 31</td>
<td>CHEM 142, 141*</td>
<td>CHEM 110*, CHEM 110L*</td>
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</tr>
<tr>
<td>LIBRARY AND INFORMATION LITERACY</td>
<td>CHEM: 1B</td>
<td>CHEM 141*</td>
<td>CHEM 110*, CHEM 164*</td>
<td></td>
</tr>
<tr>
<td>COMPUTER LITERACY</td>
<td>GE: AREA B CHEM 1A, 1B</td>
<td>CHEM 25, 31</td>
<td>CHEM 141*, CHEM 125</td>
<td>CHEM 162, CHEM 110L*</td>
</tr>
<tr>
<td>LABORATORY SKILLS</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

* Elective Option

6. ASSESSMENT GOALS

Assessment Goals

Laboratory Knowledge and Skills

Students obtaining a baccalaureate chemistry degree should have upon graduation
1. the basic analytical and technical skills to work effectively in the various fields of chemistry.
2. the ability to perform accurate quantitative measurements with an understanding of the theory and use of contemporary chemical instrumentation, interpret experimental results, perform calculations on these results and draw reasonable, accurate conclusions.
3. the ability to synthesize, separate and characterize compounds using published reactions, protocols, standard laboratory equipment, and modern instrumentation.
4. the ability to use information technology tools such as the Internet and computer-based literature searches as well as printed literature resources to locate and retrieve scientific information needed for laboratory or theoretical work.
5. the ability to present scientific and technical information resulting from laboratory experimentation in both written and oral formats.
6. knowledge and understanding of the issues of safety regulations in the use of chemicals in their laboratory work.

Assessment Goals

Computer, Library and Information Skills

Students obtaining a baccalaureate chemistry degree should have upon graduation
1. the ability to make effective use of the library and other information resources in chemistry, including
   a. finding chemical information utilizing the primary literature.
   b. critically evaluating chemical information.
   c. finding and evaluating chemical information utilizing secondary sources such as the Internet.
2. the ability to make effective use of computers in chemistry applications, including

a. using a computer as a tool in writing, drawing chemical structures and data analysis to communicate scientific information.
b. having a familiarity with the applications of computers in the modeling and simulation of chemical phenomena.
c. having an appreciation of the applications of computers in data acquisition and processing.
d. retrieval of information using library or internet resources.

Assessment Goals
Oral and Written Communication Skills in Chemistry
Students obtaining a baccalaureate chemistry degree should have upon graduation

1. adequate skills in technical writing and oral presentations. [Students must be given the opportunity to practice effective writing and oral communication throughout the chemistry curriculum.]

2. the ability to communicate scientific information in oral and written formats to both scientists and nonscientists.

Assessment Goals
Quantitative Reasoning Skills
Students obtaining a baccalaureate chemistry degree should have upon graduation

1. sufficient quantitative reasoning skills to successfully pursue their career objectives, a related career or further professional training.

2. developed their

a. proficiency in unit algebra skills.
b. ability to accurately collect and interpret numerical data.
c. ability to solve problems competently using extrapolation, approximation, precision, accuracy, rational estimation and statistical validity.
d. ability to relate theories involving numbers and the practice of the theory.
e. proficiency in the scientific method (formulating hypotheses and arriving at appropriate answers and conclusions)

Assessment Goals
Knowledge of Chemical Principles and Facts
Students obtaining a baccalaureate chemistry degree should have upon graduation

1. developed a mastery of critical thinking skills, problem-solving skills and data analysis skills leading to the ability to
a. collect and analyze data.
b. apply fundamental chemical principles to gather and explain data.
c. design experiments or model systems to test hypotheses.
d. assess the relative validity of several possible solutions to a problem.

2. a working knowledge of chemical principles appropriate to a chemistry degree program to include thermodynamics, equilibrium, kinetics, quantum mechanics, structures of materials, and synthesis.

3. a mastery of a broad set of factual chemical knowledge concerning the properties of substances, molecules and atoms.

The set of preceding assessment goals were developed and approved by CSU chemistry faculty at a May 5, 1999, statewide conference of Assessment of Chemistry Degree programs.

7. CAPSTONE PROJECTS & SENIOR RESEARCH:

Capstone projects and senior research consist of the following components in the chemistry major:

A. Preparation: Students
1) Explore a specific topic suitable for undergraduate student research or capstone laboratory project in a subdiscipline of chemistry in collaboration with one or more faculty members in the department;
2) Review safety literature and MSDS sheets of chemicals to be used in project.

B. Execution of Research or Capstone Laboratory Project
As a result of these preparative activities, the student, in collaboration with the faculty member in whose laboratory or class the student:
1) Describes the goals to be achieved in the project
2) Describes and evaluates the experimental design to be used in achieving the project goals;
3) Delineates the protocol to be followed; describes and evaluates experimental controls;
4) Identifies, describes and evaluates specific techniques and instrumentation to be used; gains competence in techniques and/or instrumentation with which he/she is not familiar;
5) Executes prescribed protocol;
6) Records data generated from the experimental protocols; provides charts, tables or graphs which summarize and/or illustrate major results; completes any descriptive and/or inferential statistical treatment of the data in order to assess the certainty associated with the results;
7) Articulates and evaluates interpretations made from the data/results;
8) Articulates and evaluates major conclusions of the experiment or study;
9) Addresses any problems encountered with achieving the project goals;
10) Proposes future experiments to improve upon or extend the present work.

C. Preparation of a Poster
(Note: Capstone courses will require a Poster. An Oral Presentation and/or written report may be required.)
The student:
1) Constructs a poster using the style and format given in the individual course guidelines. In general, the poster should closely follow the Formal Written Report.
2) Writes with clear, accurate sentence structure using appropriate terminology, spelling, grammar and punctuation.
3) Successfully communicates the research in the Poster to an audience, using the Poster as a visual aid.

D. Preparation of a Written Report
1) A written report shall conform to the appropriate research journal in the field. For example, the appropriate journal for a project in organic chemistry is *Journal of Organic Chemistry*. The instructor of the capstone course will identify the appropriate journal.
2) Generally, the report will have a title, abstract, introduction, experimental section, results, discussion and references. The format of each section is described in the respective research journal.

The student:
3) Utilizes an approved word processing program to produce hard copies of text for revision, proofreading and final copy.
4) Reviews the paper for a.) logical train of thought, b.) continuity and clarity of transitions between elements of the topic, c.) consistency of style and approach, and d.) balance between elements of the topic consistent with objective of the paper.
5) Writes with clear, accurate sentence structure using appropriate and conventional words, grammar, and punctuation; uses correct verb tense, symbols and abbreviations; avoids jargon, wordiness and repetition.

E. Preparation of an Oral Presentation

The student:
1) Utilizes a format in which the written report serves as the basis for the oral presentation and the Introduction, Materials and Methods, Results and Discussion sections comprise the major portion of the presentation;
2) Uses the appropriate audio-visual aids to illustrate the presentation in a clear and concise manner. Uses supplementary handouts during the presentation where appropriate;
3) Avoids reading the presentation; presents from a list of notes consisting of the main concepts and points to be emphasized;
4) Speaks slowly and clearly, maintaining eye contact with the entire audience; uses appropriate gestures;
5) Repeats difficult or confusing material using different wording and invites questions at points in the presentation which may be confusing;
6) Ends the presentation at the designated time limit which includes a review and emphasis of major points and conclusions of the research;

**ASSESSMENT CRITERIA FOR FORMAL LAB REPORTS**

*Format* – The report contains the required sections as per syllabus: Title, Abstract, Introduction, Results, Discussion, and References.
*Content* – Each section of the report contains the appropriate type and amount of information.
Data analysis and presentation – The data are properly analyzed as well as clearly and neatly presented.

Conclusions – The conclusions drawn from the data are consistent with the data presented. The results of the report are thoroughly discussed in the context of the present experiment with extensive reference to the existing primary literature. Errors in experimental measurements or design are thoroughly analyzed.

Writing
Grammar – The report contains good grammar, including accurate spelling, good sentence construction, appropriate use of punctuation, and proper word usage.

Organization – The writing in each section of the report is organized, with a logical flow of ideas between and within paragraphs.

Style – Each section of the report is written in the appropriate style and in passive voice.

Presentation – The report is presented well and has an overall appearance that leaves a favorable impression upon the reader.

"A" papers are those that flawlessly fulfill all of the above criteria.

"B" papers are those that generally fulfill the criteria but may be less than excellent in one or more areas. For example, a paper that has very good data presentation but a somewhat choppy or clunky writing style is more likely to be a "B" rather than an "A" paper.

"C" papers are those only fulfill some of the criteria and/or contain serious flaws such as poor sentence structure, poor paragraph organization, or shoddy data presentation.

"D" papers are those that contain major flaws in several areas.

"F" papers are those that contain major flaws in several areas and are missing critical information (such as the abstract or a reference list).

ASSESSMENT CRITERIA FOR POSTER PRESENTATIONS

Format – The poster is properly constructed in the format stated in the course guidelines and contains a Title and the required sections such as Abstract, Materials and Methods, Results, Conclusion, References.

Content – The poster successfully reports the results of the project by focusing on the essential features of the project while avoiding excessive wordiness.

Writing – The writing is smooth, concise, well-organized, with good logical flow of ideas. The writing contains good grammar with proper spelling, punctuation, sentence structure and word usage.

Results – The poster shows that the student understands the experimental procedures and experimental outcomes. The student demonstrates an understanding of the quality of the work.

Data Analysis and Presentation – The data are carefully analyzed and presented in a clean, readable format. Plots and tables are organized and clearly labeled so that the reader can grasp the essential features of the data at a glance.

Conclusions – The conclusions are clearly and simply laid out. The reader should be able to easily grasp the overall results of the study from this section alone.

Presentation – The poster presentation is clean, neat, uncongested, and easy to follow.

"A" posters are those that flawlessly fulfill all of the above criteria.

"B" posters fulfill most of the criteria. A "B" poster may contain major flaws in only one area or minor flaws in no more than two areas.

"C" posters contain major flaws in two areas or minor flaws in three areas.
"D" posters contain major flaws in three areas or minor flaws in most areas.
"F" posters contain major flaws in three areas and are missing sections or other critical information.

PRIMARY ASSESSMENT ACTIVITIES FOR THE NEXT FIVE YEARS

Alumni Survey
The first survey conducted by the department encompassed the entire alumni population. The next alumni survey is being conducted by Institutional Studies in the year 2000. A future survey of alumni will focus on alumni who graduated after the second survey was administered. The questions will focus on concerns and issues of faculty based on a review of the results of the first and second alumni survey. This survey will occur in the three years after the second survey is completed.

Writing in the Major
A random selection of laboratory/poster reports will be collected in chemistry 141 and 164 in the spring semester. A committee composed of the instructors in the two courses and two other faculty will review the reports to determine how effectively students are meeting two criteria: Organization
Technical writing skills
The committee will prepare a report to the department that discusses how effectively students performed in the two criteria. Recommendations for improving, if necessary, the performance of students will also be submitted. The faculty of the department will discuss the recommendations and adopt appropriate recommendations and action pathways.

American Chemical Society Exams
The department will initiate a study of the feasibility of using American Chemical Society examinations in the following courses
Chemistry 124
Chemistry 140B
Chemistry 160B
The first step is a study of each ACS examination by a group of faculty who teach these courses. Each study will consider: The content covered in the exams compared to the content covered in the course, the level of difficulty of the examination, how the entire exam might be administered, and security of exam issues. At the conclusion of each study, the faculty group will prepare a written report to the faculty regarding the potential for administering each exam at least once per academic year.

The purpose of administering an ACS exam is to compare the chemical knowledge competency of chemistry degree students compared to a national norm.
After receiving the reports, the faculty will determine whether to administer all or any of the ACS examinations. For those ACS examinations approved for administration, a committee of the instructors of the identified courses will meet annually and review the results. A report will be developed and submitted to the department for its review. At the end of the trial period, the department will determine if continuation of the exams is justified.

ASSESSmission/06/19/00
<table>
<thead>
<tr>
<th>Learning Outcome</th>
<th>Courses assessed</th>
<th>Year and frequency of assessment</th>
<th>Assessment activity</th>
<th>Evaluation</th>
<th>Data collection</th>
<th>Analysis, reporting, and follow-through</th>
<th>BLG</th>
</tr>
</thead>
<tbody>
<tr>
<td>A. Laboratory Knowledge and Skills</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
<td>Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.</td>
<td>19</td>
</tr>
<tr>
<td>1. the basic analytical and technical skills to work effectively in the various fields of chemistry</td>
<td></td>
<td></td>
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<td></td>
</tr>
<tr>
<td>2. the ability to perform accurate quantitative measurements with an understanding of the theory and use of contemporary chemical instrumentation, interpret experimental results, perform calculations on these results and draw reasonable, accurate conclusions.</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
<td>Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.</td>
<td>5, 16</td>
</tr>
<tr>
<td>3. the ability to synthesize, separate and characterize compounds using published reactions, protocols, standard laboratory equipment, and modern instrumentation.</td>
<td>Not assessed at program level</td>
<td></td>
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<tr>
<td>4. the ability to use information technology tools such as the Internet and computer-based literature searches as well as printed literature resources to locate and retrieve</td>
<td>110L, 125, 141, 164</td>
<td>Every semester offered</td>
<td>Capstone poster project</td>
<td>Poster project rubric</td>
<td>Multiple faculty evaluation during department poster session</td>
<td>Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.</td>
<td>2, 19</td>
</tr>
</tbody>
</table>
### 5. the ability to present scientific and technical information resulting from laboratory experimentation in both written and oral formats.

<table>
<thead>
<tr>
<th>Course</th>
<th>Every semester offered</th>
<th>Capstone poster project</th>
<th>Poster project rubric</th>
<th>Multiple faculty evaluation during departmen't poster session</th>
<th>Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.</th>
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<tr>
<td>110L, 125, 141, 164</td>
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</table>

### 6. knowledge and understanding of the issues of safety regulations in the use of chemicals in their laboratory work.

<table>
<thead>
<tr>
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<th>Not assessed at program level</th>
<th>N/A</th>
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### B. Computer, Library and Information Skills

#### 1. the ability to make effective use of the library and other information resources in chemistry, including the primary literature, tabulated data, and secondary sources such as the Internet.

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#### 2. the ability to make effective use of computers in chemistry applications using standard and chemistry specific software packages.

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#### 3. the ability to perform and interpret simple molecular modeling or chemical computations using standard software.

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### C. Oral and Written Communication
### n Skills in Chemistry

1. **adequate skills in technical writing and oral presentations.**
   - **Courses:** 110L, 125, 141, 164
   - **Offered:** Every semester
   - **Poster Project:** Multiple faculty evaluation during department poster session
   - **Evaluation:** Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.

2. **the ability to communicate scientific information in oral and written formats to both scientists and nonscientists.**
   - **Courses:** 110L, 125, 141, 164
   - **Offered:** Every semester
   - **Poster Project:** Multiple faculty evaluation during department poster session
   - **Evaluation:** Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.

### D. Quantitative Reasoning Skills

1. **ability to accurately collect and interpret numerical data.**
   - **Courses:** 110L, 125, 141, 164
   - **Offered:** Every semester
   - **Poster Project:** Multiple faculty evaluation during department poster session
   - **Evaluation:** Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.

2. **ability to solve problems competently using extrapolation, approximation, precision, accuracy, rational estimation and statistical validity.**
   - **Courses:** 110L, 141
   - **Offered:** Every semester
   - **Poster Project:** Multiple faculty evaluation during department poster session
   - **Evaluation:** Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.

3. **proficiency in the scientific method (formulating hypotheses and arriving at appropriate answers and conclusions)**
   - **Courses:** 110L, 125, 141, 164
   - **Offered:** Every semester
   - **Poster Project:** Multiple faculty evaluation during department poster session
   - **Evaluation:** Department chair prepares aggregate data and evaluation for department discussion; submission to Office of Assessment.

### E. Knowledge of Chemical Principles and Facts

1. **a working knowledge of chemical principles appropriate to a chemistry degree program to include thermodynamics, equilibrium, kinetics, quantum**
   - **Assessment:** ACS Standardized Exam
   - **Completed:** in 110, 124, 160B every semester
   - **Evaluation:** Course instructor
   - **Scores:** Scores are compared to national norms. Instructors evaluate performance in different areas of exam and adjust lectures accordingly. Department chair prepares aggregate and historical data for submission to Office of Assessment.
| 2. a mastery of a broad set of factual chemical knowledge concerning the properties of substances, molecules, and atoms. | 110, 24/124, 160A/B | Assessment conducted in 110, 124, 160B every semester offered | ACS Standardized Exam | ACS Exzmk key | Course instructor | Scores are compared to national norms. Instructors evaluate performance in different areas of exam and adjust lectures accordingly. Department chair prepares aggregate and historical data for submission to Office of Assessment. | 18 |