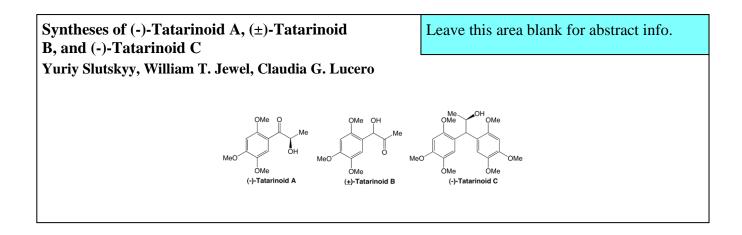
# **Graphical Abstract**





Tetrahedron Letters

journal homepage: www.elsevier.com

## Syntheses of (-)-Tatarinoid A, (±)-Tatarinoid B, and (-)-Tatarinoid C

Yuriy Slutskyy, William T. Jewel<sup>a</sup>, Claudia G. Lucero<sup>b,</sup>\*

<sup>a</sup>University of California, Davis, One Shields Ave., Davis, CA, 95616, USA <sup>b</sup>Department of Chemistry, California State University, Sacramento, 6000 J Street, Sacramento, CA, 95819, USA

### ARTICLE INFO

ABSTRACT

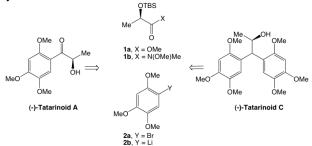
Article history: Received Received in revised form Accepted Available online

Keywords: (-)-Tatarinoid A (±)-Tatarinoid B (-)-Tatarinoid C Irregular Wittig reaction The syntheses of (–)-Tatarinoid A, ( $\pm$ )-Tatarinoid B, and (–)-Tatarinoid C in 1 to 3 steps are described herein. (–)-Tatarinoid A and (–)-Tatarinoid C are both constructed in 3 steps from 1-bromo-2,4,5-trimethoxybenzene in overall yields of 63% and 74%, respectively. The addition of (1-methoxyethyl)triphenylphosphonium ylide to 2,4,5-trimethoxybenzaldehyde provides ( $\pm$ )-Tatarinoid B in 1 step in 97% yield.

2012 Elsevier Ltd. All rights reserved.

(–)-Tatarinoids A, B, and C are 3 of 19 compounds that have been isolated from the rhizome of the plant *Acorus tatarinowii*.<sup>1</sup> Used in Chinese medicine, *Acorus tatarinowii* possesses pharmacological effects on the central nervous system by regulating cyclic adenosine monophosphate (cAMP) activity. Although (–)-Tatarinoids B and C have shown weak efficacy we continued our efforts towards the syntheses of these natural products since currently there are no reported syntheses of (–)-Tatarinoids A, C and (±)-Tatarinoid B. Herein we report expeditious routes towards these natural products.

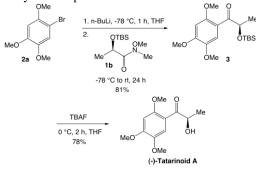
As illustrated in Scheme 1, (–)-Tatarinoid A, and Tatarinoid C can be accessed from 1-bromo-2,4,5-trimethoxybenzene (2a). The required organolithium 2b would be accessed through a metal halogen exchange between the aryl bromide (2a) and *n*-butyllithium.



Scheme 1. Retrosynthesis of (-)-Tatarinoid A and (-)-Tatarinoid C

In the case of (–)-Tatarinoid A it would be necessary to convert the TBS protected methyl (R)–lactate (1a) to the Weinreb amide (1b) in order to avoid the over addition of the aryllithium (2b) to the carbonyl and yield only the necessary ketone, a precursor to the natural product.<sup>2</sup>

Alternatively, the desired (–)-Tatarinoid C would employ the TBS protected methyl (R)–lactate (1a)<sup>3</sup>, in order to allow for the over addition of the aryllithium (2b) and result in the formation of the tertiary alcohol precursor.



Scheme 2. Synthesis of (-)-Tatarinoid A

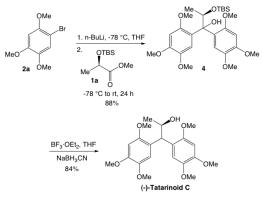
The synthesis of (–)-Tatarinoid A was shown to proceed in three steps from Weinreb amide  $1b^{4,5}$  and aryl bromide **2a** (Scheme 2).<sup>6</sup> Formation of the aryllithium (**2b**) proceeded from the treatment of the aryl bromide (**2a**) with *n*-BuLi at —78 °C for

<sup>\*</sup> Corresponding author. Tel.: 1-916-397-8100; fax: 1-916-278-4986; *E-mail address*: cglucero@csus.edu

1 hour. Initial investigations towards the total synthesis of (-)-Tatarinoid A included the addition of aryllithium to the TBS protected ester **1a**. The mono addition product, ketone **3**, was observed as a minor product. As expected, the over addition of the aryllithium to yield the tertiary alcohol (**4**), the precursor to (-)-Tatarinoid C (Scheme 3), was the major product.

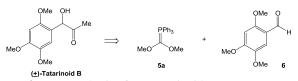
To avoid the formation of the tertiary alcohol, we turned our attention towards the derivatization of the TBS protected ester **1a** to the Weinreb amide **1b**, synthesized from the treatment of ester **1a** with *N*,*O*-dimethylhydroxylamine hydrochloride salt and *i*-PrMgCl in THF. The reaction proceeded smoothly in 99% yield without purification. As predicted, the mono addition of the aryllithium to the amide resulted in the desired ketone, **3**, in a 81% yield, without any of the overaddition product observed. Finally, removal of the TBS protecting group with TBAF (tetrabutylammonium fluoride)<sup>7</sup> in THF, at 0 °C for 2 hours, afforded (–)-Tatarinoid A in 78% yield. Assembly of the natural product was accomplished in three steps and with an overall yield of 63%.

The synthesis of (-)-Tatarinoid C employed a similar strategy to that of (-)-Tatarinoid A. In this case, the over addition of the aryllithium (2b) to the TBS protected ester (2a) was crucial to provide the desired tertiary alcohol 4 (Scheme 3). The reaction was accomplished using 3 equivalents of both aryl bromide and *n*-BuLi to yield **4** in 88% yield. Treatment of the tertiary alcohol with BF<sub>3</sub>·OEt<sub>2</sub> and NaBH<sub>3</sub>CN at room temperature for 1 hour was expected to only afford the desired deoxygenated product, which would then be treated with TBAF for the removal of the protecting group. However, the reaction afforded (-)-Tatarinoid C in 84% yield, suggesting that the excess NaBH<sub>3</sub>CN (3 equiv.) permits for the reductive cleavage of the TBS group.<sup>8</sup> The resulting natural product, (-)-Tatarinoid C, was completed in three steps from 1-bromo-2,4,5-trimethoxybenzene (2a) and methyl (R)-2-(tert-butyldimethylsilyloxy)propionate (1a) with an overall yield of 74%.



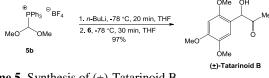
Scheme 3. Synthesis of (-)-Tatarinoid C

A retrosynthetic overview of  $(\pm)$ -Tatarinoid B is illustrated in Scheme 4. It was expected that the racemic mixture of the natural product would be accessed in one step from an irregular Wittig reaction using ylide (**5a**) and 2,4,5trimethoxybenzaldehyde (**6**).<sup>10,11</sup>



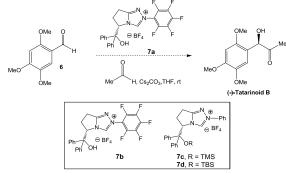
Scheme 4. Retrosynthesis of (±)-Tatarinoid B

Indeed, treatment of (1-methoxyethyl)triphenylphosphonium salt (**5b**) with *n*-BuLi, for 20 minutes, provided the desired Wittig ylide **5a** that was then treated with 2,4,5-trimethoxybenzaldehyde (**6**) at -78 °C for 30 minutes to yield the racemic Tatarinoid B in 97% yield. No sign of the typical Wittig product was observed.



Scheme 5. Synthesis of (±)-Tatarinoid B

Currently, we are turning our efforts toward the construction of (-)-Tatarinoid B in one step from a crossed acyloin condensation<sup>12</sup> between 2,4,5-trimethoxybenzaldehyde (**6**) and acetaldehyde using a chiral triazolium salt as a catalyst (Scheme 6, **7a-7d**).<sup>13,14</sup> These catalysts are not commercially available and are themselves constructed in five to six steps from methyl L-pyroglutamate. Our continued investigation on the total synthesis of (-)-Tatarinoid B, high yield and enantioselectivity, will be reported in due course.



Scheme 6. Synthesis of (-)-Tatarinoid B

In conlusion, we have described the first total syntheses of (–)-Tatarinoid A and C and racemic Tatarinoid B. The synthesis of (–)-Tatarinoid A was accomplished in three steps with an overall yield of 63%. ( $\pm$ )-Tatarinoid B was constructed in one step in 97% yield while (–)-Tatarinoid C was synthesized also in three steps with an overall yield of 74%.

#### Acknowledgments

The authors would like to thank CSUPERB Presidents' Commission Scholars Program for funding Yuriy Slutskyy. Thanks are also due to Erich Bowman, Maddy McCrea-Hendrick, Tyler Johnson, and Olga Inozemtseva for their contributions to the project.

#### **References and notes**

- Tong, X.-G.; Wu, G.-S.; Huang, C.-G.; Lu, Q.; Wang, Y.-H.; Long, C.-L.; Luo, H.-R.; Zhu, H.-J.; Cheng, Y.-X. J. Nat. Prod. 2010, 73, 1160-1163.
- 2. Marcus, A. P.; Sarpong, R. Org. Lett. 2010, 12, 4560-4563.
- Sakai, A.; Aoyama, T.; Shioiri, T.; *Tetrahedron Lett.* 2000, 6859-6863.
  Mandal, A. K.; Schneekloth, J. S.; Crews, C. M. *Org. Lett.* 2005, 7,
- 3645-3648.
  Nahm, tS.; Weinreb, S. M. *Tetrahedron Lett.*1981, 22, 3815-3818.
- Sutherland, H. S.; Higg, K. C.; Taylor, N. J. *Tetrahedron*, 2001, 57,
- 309-317.
- Huang, S.-X.; Powell, E.; Rajski, S. R.; Zhao, L.-X.; Jiang, C.-L.; Duan, Y.; Xu, W.; Shen, B. Org. Lett. 2010, 12, 3225-3527.
- 8. Srikrishna, A.; Viswajanani, R.; Sattigeri, J. A.; Yelamaggad, C. V. *Tetrahedron Lett.* **1995**, *36*, 2347-2350.
- 9. Corey, E. J.; Jones, G. B. J. Org. Chem. 1992, 57, 1028-1029.

- Okada, H.; Mori, T.; Saikawa, Y.; Nakata, M. *Tetrahedron Lett.* 2009, 50, 1276-1278.
- 11. Lin, Y.-L.; Wu, C.-S.; Lin, S.-W.; Huang, J.-L.; Sun, Y.-S.; Yand, D.-Y. *Bioorganic & Medicinal Chem.* **2002**, *10*, 685-690.
- 12. Jin, M. Y.; Kin, S. M.; Han, H.; Ryu, D. H.; Yang, J. W. Org. Lett. **2011**, *13*, 880-883.
- (a) Kerr, M.; Read de Alaniz, J.; Rovis, T. *J. Org. Chem.* 2005, *70*, 5725-5728. (b) Vora, H. U.; Lathrop, S. P.; Reynolds, N. T.; Kerr, M.; Read de Alaniz, J.; Rovis, T. *Org. Syn.* 2010, *87*, 350-357.
- (a) Marigo, M.; Wabnitz, T. C.; Fielenbach, D.; Jorgensen, K. A. *Angew. Chem. Int. Ed.* **2005**, *44*, 794-797. (b) Ostendorf, M.; Dijkink, J.; Rutjes, F. P. J. T.; Hiemstra, H. *Eur. J. Org. Chem.* **2000**, 115-124. (c) Zhang, Y.-R.; He, L.; Wu, X.; Shao, P.-L.; Ye, S. *Org. Lett.* **2008**, *10*, 277-280.

#### **Supplementary Material**

Supplementary data associated with this article may be found, in the online version, at.