

Compound Name	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
manganese (II) bromite		
manganese (II) phosphite		
rubidium sulfite		
hydroselenic acid		
sodium perbromate		
cobalt (III) chromate		
antimony (V) nitrite		
chloric acid		
pentaselenium decabromide		
disulfur decachloride		
nickel (III) nitrate		
copper (II) bromide		
nickel (II) hydrogen phosphate		
iron (II) hydrogen sulfate		
bismuth (V) acetate		
sulfurous acid		
sulfuric acid		
nickel (II) chloride		
tin (IV) phosphate		
mercury (I) iodate		

Compound Formula	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
Co(HCO ₃) ₂		
CS ₂ S		
Ca(IO ₂) ₂		
Ba ₂ C		
Mn(CO ₃) ₂		
CuBrO ₂		
AgHS		
C ₉ N ₁₀		
CrI ₂		
Mg(NO ₃) ₂		
HC ₂ H ₃ O ₂ (aq)		
HClO ₂ (aq)		
Be(IO ₄) ₂		
HIO ₄ (aq)		
BaO		
Cd(BrO ₃) ₂		
Bi(CN) ₅		
AuHS		
AuClO		
Na ₂ CO ₃		

Compound Name	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
bismuth (III) phosphide		
antimony (V) oxide		
ammonium carbonate		
nitrous acid		
barium fluoride		
iron (II) hydrogen sulfate		
magnesium nitrite		
beryllium iodate		
cadmium chromate		
gold (I) hydrogen phosphate		
bismuth (III) hydrogen sulfide		
cesium oxalate		
tin (II) iodite		
beryllium acetate		
tin (II) sulfate		
antimony (V) carbide		
cobalt (III) hydride		
sodium carbide		
dinitrogen triselenide		
potassium hypoiodite		

Compound Formula	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
$\text{Sb}_3(\text{BO}_3)_5$		
NiF_3		
C_8O_3		
$\text{HBrO}_2 \text{ (aq)}$		
$\text{Bi}(\text{ClO}_2)_3$		
$\text{H}_2\text{C}_2\text{O}_4\text{(aq)}$		
O_6F_{10}		
$\text{HC}_2\text{H}_3\text{O}_2\text{(aq)}$		
$\text{Si}_{10}\text{As}_5$		
N_2O_7		
Cl_4O_3		
Hg_2Se		
Li_4C		
CrC_2O_4		
$\text{Ni}_3(\text{AsO}_4)_3$		
$\text{Mg}(\text{ClO}_4)_2$		
Au_3BO_3		
$\text{Zn}(\text{IO})_2$		
I_5Cl_8		
$\text{Mn}(\text{HCO}_3)_2$		

Compound Name	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
cobalt (III) carbide		
aluminum sulfate		
diphosphorous octaoxide		
manganese (II) phosphate		
selenic acid		
hexaiodine nonanitride		
nitric acid		
mercury (I) nitride		
aluminum perchlorate		
chromic acid		
lithium hydrogen sulfide		
cobalt (II) sulfate		
cesium sulfate		
manganese (II) chromate		
lead (II) hydride		
ammonium nitrate		
sodium chlorate		
radium hydrogen carbonate		
copper (I) hydrogen sulfate		
arsenic (V) nitrate		

Compound Formula	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
Pb(HPO ₄) ₂		
Sn(BrO ₄) ₂		
Si ₃ As ₁₀		
N ₁₀ O ₁₀		
HBrO _(aq)		
Fe(OH) ₃		
I ₅ F ₃		
H ₂ SO _{3(aq)}		
ZnHPO ₄		
Au ₂ SO ₃		
Hg(IO ₄) ₂		
Ra(BrO ₃) ₂		
MnSO ₃		
HIO _{3 (aq)}		
CBr		
Mn(BrO) ₂		
S ₂ O ₇		
BeCrO ₄		
HBrO _{3(aq)}		
SrS		

Compound Name	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
bismuth (III) selenide		
nickel (III) hydrogen phosphate		
nonanitrogen tetroxide		
lithium hypochlorite		
cobalt (III) cyanide		
hydroselenic acid		
manganese (IV) hydrogen phosphate		
copper (II) borate		
iron (III) sulfate		
nickel (III) phosphite		
hydroiodic acid		
hexasilicon heptoxide		
pentaarsenic triphosphide		
hexacarbon trioxide		
antimony (V) hypoiodite		
sodium hypobromite		
sulfurous acid		
lithium oxalate		
aluminum chlorite		
hydrofluoric acid		

Compound Formula	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
HMnO _{4(aq)}		
AuHCO ₃		
KF		
CrPO ₄		
Ca(ClO ₂) ₂		
Ni ₂ O ₃		
SeC ₉		
Fe(IO ₄) ₃		
PO ₄		
SCl ₁₀		
BaO		
Au(IO ₃) ₃		
H ₃ PO _{4(aq)}		
Cs ₂ CrO ₄		
SnF ₄		
Ni(ClO ₂) ₂		
Al ₂ (SO ₃) ₃		
FeO		
Hg ₂ (MnO ₄) ₂		
Cl ₁₀ F ₅		

Compound Name	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
antimony (III) bromide		
hydroiodic acid		
lithium fluoride		
octaarsenic trisulfide		
copper (I) phosphite		
nickel (III) hydride		
titanium (III) hydrogen carbonate		
arsenic acid		
chromium (III) nitride		
strontium hydrogen sulfate		
magnesium hydroxide		
antimony (V) acetate		
cobalt (III) chromate		
zinc hydrogen sulfite		
copper (II) bromide		
nickel (II) hydrogen sulfide		
magnesium bromite		
antimony (V) selenate		
mercury (II) sulfite		
phosphorous acid		

Compound Formula	Indicate Type of Compound: I = ionic, A= acid, M = molecular	Write your answer here
Mn ₃ N ₄		
Ni(IO) ₃		
Sn(ClO) ₄		
Sb(HSO ₃) ₃		
CO ₃ ⁻²		
Al(MnO ₄) ₃		
Li ₂ CrO ₄		
FeP		
KHSO ₄		
HNO _{2(aq)}		
SnCO ₃		
Cl ₉ O ₇		
FeBO ₃		
H ₂ Se _(aq)		
Fe(NO ₃) ₂		
ZnSO ₄		
H ₂ S _(aq)		
SeN ₂		
BaSO ₃		
Bi ₃ P ₅		