

Compound Name	Indicate Type of Compound: I = ionic, A = acid, M = molecular	Write your answer here
manganese (II) bromite	I	$\text{Mn}(\text{BrO}_2)_2$
manganese (II) phosphite	I	$\text{Mn}_3(\text{PO}_3)_2$
rubidium sulfite	I	Rb_2SO_3
hydroselenic acid	A	$\text{H}_2\text{Se}_{(\text{aq})}$
sodium perbromate	I	NaBrO_4
cobalt (III) chromate	I	$\text{Co}_2(\text{CrO}_4)_3$
antimony (V) nitrite	I	$\text{Sb}(\text{NO}_2)_5$
chloric acid	A	$\text{HClO}_{3(\text{aq})}$
pentaselenium decabromide	M	$\text{Se}_5\text{Br}_{10}$
disulfur decachloride	M	S_2Cl_{10}
nickel (III) nitrate	I	$\text{Ni}(\text{NO}_3)_3$
copper (II) bromide	I	CuBr_2
nickel (II) hydrogen phosphate	I	NiHPO_4
iron (II) hydrogen sulfate	I	$\text{Fe}(\text{HSO}_4)_2$
bismuth (V) acetate	I	$\text{Bi}(\text{C}_2\text{H}_3\text{O}_2)_5$
sulfurous acid	A	$\text{H}_2\text{SO}_{3(\text{aq})}$
sulfuric acid	A	$\text{H}_2\text{SO}_{4(\text{aq})}$
nickel (II) chloride	I	NiCl_2
tin (IV) phosphate	I	$\text{Sn}_3(\text{PO}_4)_4$
mercury (I) iodate	I	$\text{Hg}_2(\text{IO}_3)_2$

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Co(HCO ₃) ₂	I (with VOS metal)	cobalt (II) hydrogen carbonate
Cs ₂ S	I	cesium sulfide
Ca(IO ₂) ₂	I	calcium iodite
Ba ₂ C	I	barium carbide
Mn(CO ₃) ₂	I (with VOS metal)	manganese (IV) carbonate
CuBrO ₂	I (with VOS metal)	copper (I) bromite
AgHS	I	silver hydrogen sulfide
C ₉ N ₁₀	M	nonacarbon decanitride
CrI ₂	I (with VOS metal)	chromium (II) iodide
Mg(NO ₃) ₂	I	magnesium nitrate
HC ₂ H ₃ O ₂ (aq)	A	acetic acid
HClO ₂ (aq)	A	chlorous acid
Be(IO ₄) ₂	I	beryllium periodate
HIO _{4(aq)}	A	periodic acid
BaO	I	barium oxide
Cd(BrO ₃) ₂	I	cadmium bromate
Bi(CN) ₅	I (with VOS metal)	bismuth (V) cyanide
AuHS	I (with VOS metal)	gold (I) hydrogen sulfide
AuClO	I (with VOS metal)	gold (I) hypochlorite
Na ₂ CO ₃	I	sodium carbonate

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bismuth (III) phosphide	I	BiP
antimony (V) oxide	I	Sb ₂ O ₅
ammonium carbonate	I	(NH ₄) ₂ CO ₃
nitrous acid	A	HNO _{2(aq)}
barium fluoride	I	BaF ₂
iron (II) hydrogen sulfate	I	Fe(HSO ₄) ₂
magnesium nitrite	I	Mg(NO ₂) ₂
beryllium iodate	I	Be(IO ₃) ₂
cadmium chromate	I	CdCrO ₄
gold (I) hydrogen phosphate	I	Au ₂ HPO ₄
bismuth (III) hydrogen sulfide	I	Bi(HS) ₃
cesium oxalate	I	Cs ₂ C ₂ O ₄
tin (II) iodite	I	Sn(IO ₂) ₂
beryllium acetate	I	Be(C ₂ H ₃ O ₂) ₂
tin (II) sulfate	I	SnSO ₄
antimony (V) carbide	I	Sb ₄ C ₅
cobalt (III) hydride	I	CoH ₃
sodium carbide	I	Na ₄ C
dinitrogen triselenide	M	N ₂ Se ₃
potassium hypoiodite	I	KIO

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Sb ₃ (BO ₃) ₅	I (with VOS metal)	antimony (V) borate
NiF ₃	I (with VOS metal)	nickel (III) fluoride
C ₈ O ₃	M	octacarbon trioxide
HBrO _{2(aq)}	A	bromous acid
Bi(ClO ₂) ₃	I (with VOS metal)	bismuth (III) chlorite
H ₂ C ₂ O _{4(aq)}	A	oxalic acid
O ₆ F ₁₀	M	hexooxygen decafluoride
HC ₂ H ₃ O _{2(aq)}	A	acetic acid
Si ₁₀ As ₅	M	decasilicon pentaarsenide
N ₂ O ₇	M	dinitrogen heptoxide
Cl ₄ O ₃	M	tetrachlorine trioxide
Hg ₂ Se	I (with VOS metal)	mercury (I) selenide
Li ₄ C	I	lithium carbide
CrC ₂ O ₄	I (with VOS metal)	chromium (II) oxalate
NiAsO ₄	I (with VOS metal)	nickel (III) arsenate
Mg(ClO ₄) ₂	I	magnesium perchlorate
Au ₃ BO ₃	I (with VOS metal)	gold (I) borate
Zn(IO) ₂	I	zinc hypoiodite
I ₅ Cl ₈	M	pentaiodine octachloride
Mn(HCO ₃) ₂	I (with VOS metal)	manganese (II) hydrogen carbonate

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cobalt (III) carbide	I	Co ₄ C ₃
aluminum sulfate	I	Al ₂ (SO ₄) ₃
diphosphorous octaoxide	M	P ₂ O ₈
manganese (II) phosphate	I	Mn ₃ (PO ₄) ₂
selenic acid	A	H ₂ SeO _{4(aq)}
hexaiodine nonanitride	M	I ₆ N ₉
nitric acid	A	HNO _{3(aq)}
mercury (I) nitride	I	(Hg ₂) ₃ N ₂
aluminum perchlorate	I	Al(ClO ₄) ₃
chromic acid	A	H ₂ CrO _{4(aq)}
lithium hydrogen sulfide	I	LiHS
cobalt (II) sulfate	I	CoSO ₄
cesium sulfate	I	Cs ₂ SO ₄
manganese (II) chromate	I	MnCrO ₄
lead (II) hydride	I	PbH ₂
ammonium nitrate	I	NH ₄ NO ₃
sodium chlorate	I	NaClO ₃
radium hydrogen carbonate	I	Ra(HCO ₃) ₂
copper (I) hydrogen sulfate	I	CuHSO ₄
arsenic (V) nitrate	I	As(NO ₃) ₅

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Pb(HPO ₄) ₂	I (with VOS metal)	lead (IV) hydrogen phosphate
Sn(BrO ₄) ₂	I (with VOS metal)	tin (II) perbromate
Si ₃ As ₁₀	M	trisilicon decaarsenide
N ₁₀ O ₁₀	M	decanitrogen decoxide
HBrO _(aq)	A	hypobromous acid
Fe(OH) ₃	I (with VOS metal)	iron (III) hydroxide
I ₅ F ₃	M	pentaiodine trifluoride
H ₂ SO _{3(aq)}	A	sulfurous acid
ZnHPO ₄	I	zinc hydrogen phosphate
Au ₂ SO ₃	I (with VOS metal)	gold (I) sulfite
Hg(IO ₄) ₂	I (with VOS metal)	mercury (II) periodate
Ra(BrO ₃) ₂	I	radium bromate
MnSO ₃	I (with VOS metal)	manganese (II) sulfite
HIO _{3 (aq)}	I	iodic acid
CBr	M	carbon bromide
Mn(BrO) ₂	I (with VOS metal)	manganese (II) hypobromite
S ₂ O ₇	M	disulfur heptoxide
BeCrO ₄	I	beryllium chromate
HBrO _{3(aq)}	A	bromic acid
SrS	I	strontium sulfide

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bismuth (III) selenide	I	Bi ₂ Se ₃
nickel (III) hydrogen phosphate	I	Ni ₂ (HPO ₄) ₃
nonanitrogen tetroxide	M	N ₉ O ₄
lithium hypochlorite	I	LiClO
cobalt (III) cyanide	I	Co(CN) ₃
hydroselenic acid	A	H ₂ Se _(aq)
manganese (IV) hydrogen phosphate	I	Mn(HPO ₄) ₂
copper (II) borate	I	Cu ₃ (BO ₃) ₂
iron (III) sulfate	I	Fe ₂ (SO ₄) ₃
nickel (III) phosphite	I	NiPO ₃
hydroiodic acid	A	HI _(aq)
hexasilicon heptoxide	M	Si ₆ O ₇
pentaarsenic triphosphide	M	As ₅ P ₃
hexacarbon trioxide	M	C ₆ O ₃
antimony (V) hypoiodite	I	Sb(IO) ₅
sodium hypobromite	I	NaBrO
sulfurous acid	A	H ₂ SO _{3(aq)}
lithium oxalate	I	Li ₂ C ₂ O ₄
aluminum chlorite	I	Al(ClO ₂) ₃
hydrofluoric acid	A	HF _(aq)

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$\text{HMnO}_4\text{(aq)}$	A	permanganic acid
AuHCO_3	I (with VOS metal)	gold (I) hydrogen carbonate
KF	I	potassium fluoride
CrPO_4	I (with VOS metal)	chromium (III) phosphate
$\text{Ca}(\text{ClO}_2)_2$	I	calcium chlorite
Ni_2O_3	I (with VOS metal)	nickel (III) oxide
SeC_9	M	selenium nonacarbide
$\text{Fe}(\text{IO}_4)_3$	I (with VOS metal)	iron (III) periodate
PO_4	M	phosphorous tetroxide
SCl_{10}	M	sulfur decachloride
BaO	I	barium oxide
$\text{Au}(\text{IO}_3)_3$	I (with VOS metal)	gold (III) iodate
$\text{H}_3\text{PO}_4\text{(aq)}$	A	phosphoric acid
Cs_2CrO_4	I	cesium chromate
SnF_4	I (with VOS metal)	tin (IV) fluoride
$\text{Ni}(\text{ClO}_2)_2$	I (with VOS metal)	nickel (II) chlorite
$\text{Al}_2(\text{SO}_3)_3$	I	arsenic (V) sulfite
FeO	I (with VOS metal)	iron (II) oxide
$\text{Hg}_2(\text{MnO}_4)_2$	I (with VOS metal)	mercury (I) permanganate
Cl_{10}F_5	M	decachlorine pentafluoride

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antimony (III) bromide	I	SbBr_3
hydroiodic acid	A	$\text{HI}_{(\text{aq})}$
lithium fluoride	I	LiF
octaarsenic trisulfide	M	As_8S_3
copper (I) phosphite	I	Cu_3PO_3
nickel (III) hydride	I	NiH_3
titanium (III) hydrogen carbonate	I	$\text{Ti}(\text{HCO}_3)_3$
arsenic acid	I	$\text{H}_3\text{AsO}_{4(\text{aq})}$
chromium (III) nitride	I	CrN
strontium hydrogen sulfate	I	$\text{Sr}(\text{HSO}_4)_2$
magnesium hydroxide	I	$\text{Mg}(\text{OH})_2$
antimony (V) acetate	I	$\text{Sb}(\text{C}_2\text{H}_3\text{O}_2)_5$
cobalt (III) chromate	I	$\text{Co}_2(\text{CrO}_4)_3$
zinc hydrogen sulfite	I	$\text{Zn}(\text{HSO}_3)_2$
copper (II) bromide	I	CuBr_2
nickel (II) hydrogen sulfide	I	$\text{Ni}(\text{HS})_2$
magnesium bromite	I	$\text{Mg}(\text{BrO}_2)_2$
antimony (V) selenate	I	$\text{Sb}_2(\text{SeO}_4)_5$
mercury (II) sulfite	I	HgSO_3
phosphorous acid	A	$\text{H}_3\text{PO}_{3(\text{aq})}$

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Mn ₃ N ₄	I (with VOS metal)	manganese (IV) nitride
Ni(IO) ₃	I (with VOS metal)	
Sn(ClO) ₄	I (with VOS metal)	tin (IV) hypochlorite
Sb(HSO ₃) ₃	I (with VOS metal)	antimony (III) hydrogen sulfite
CO ₃ ⁻²	this is an ION not a compound!	carbonate
Al(MnO ₄) ₃	I	aluminum permanganate
Li ₂ CrO ₄	I	lithium chromate
FeP	I (with VOS metal)	iron (III) phosphide
KHSO ₄	I	potassium hydrogen sulfate
HNO _{2(aq)}	A	nitrous acid
SnCO ₃	I (with VOS metal)	tin (II) carbonate
Cl ₉ O ₇	M	nonachlorine heptoxide
FeBO ₃	I (with VOS metal)	iron (III) borate
H ₂ Se _(aq)	A	hydroselenic acid
Fe(NO ₃) ₂	I (with VOS metal)	iron (II) nitrate
ZnSO ₄	I	zinc sulfate
H ₂ S _(aq)	A	hydrosulfuric acid
SeN ₂	M	selenium dinitride
BaSO ₃	I	barium sulfite
Bi ₃ P ₅	I (with VOS metal)	bismuth (V) phosphide